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高含水油包水乳液的水合物储气性能研究

于

510640

Span80 / Tween80 M_{Tween80} $M_{\text{Span80}}=0.783$ 1 5 wt%

274.15 K 10:1 6 MPa 700
r/min 141.42 L /L
130 L /L

TE 82 A

Investigation on hydrate-based methane storage properties in
water-in-oil emulsion with high water content

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Abstract: Water-in-oil emulsion is a newly emerging hydration enhancement material in recent years. It has good hydration and gas storage potential. However, in order to ensure the stability of emulsion, the water content of the water-in-oil emulsion usually does not exceed 50%. The gas storage capacity of hydrate is closely related to the water content. Therefore, high water content water-in-oil emulsion has more application prospects. In this study, a water-in-oil emulsion with water content of 55% was produced and the methane hydrate formation experiment was carried out by using the emulsion. The effects of the amount of emulsifier, initial pressure, stirring rate and cyclic capacity were investigated. The results show that the increase of water content will lead to the increase of emulsion droplet and the decrease of gas storage when the water content is more than 55%. The water content of the emulsion is 55% with 5 wt% (based on water content) of Span80 / Tween80 (M_{Tween80} $M_{\text{Span80}}=0.783$ 1) emulsion is the best emulsion for storing CH₄. The increase of initial pressure is beneficial to the improvement of hydrated gas storage performance. However, too high pressure will lead to the rapid formation of hydrate shell and thus reduce the overall gas storage capacity. The appropriate stirring rate is conducive to hydrate formation. The optimal emulsion hydration and gas storage conditions were temperature of 274.15K, gas-water volume ratio in the reactor of 10:1, methane initial pressure of 6 MPa, and stirring rate of 700 r/min. In this condition, the gas storage capacity was 141.42 L_{gas}/L_{water}; further four-cyclic gas storage experiment proved that the emulsion had good recyclability, which cyclic gas storage capacity is above 130 L_{gas}/L_{water}. The research results can provide technical supporting for natural gas storage and transportation and hydrocarbon-containing mixed gas separation.

Key words: methane; hydrate; emulsion; hydration rate; cyclic gas storage

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1979-

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2016YFC0304006 2017YFC0307302

2018A030313245

(51876069 21736005)

于
 34970A
 ±0.1 K
 253~373K
 Pt-100
 40 MPa
 0-25 MPa
 THD-3030
 ±0.1K
 ±0.25%

1

Fig. 1 Schematic diagram of static methane hydration experimental device

2

XT5405FSC

243~323K

2

Fig. 2 Schematic diagram of stirred methane hydration experimental device

1.3

Span80

Tween80

MPa

298.15K

1.4

1.4.1 水合储气量

1

$$C_v = \frac{n_{CH_4}^H(t)}{V_{H_2O}} \times 22400 \quad \text{mL/mol} \quad (\text{L} / \text{L})$$

1

$$C_v \qquad \qquad \qquad \text{L} / \text{L} \qquad \qquad \qquad V_{H_2O} \qquad \qquad \qquad \text{mL} \quad n_{CH_4}^H(t) \qquad \qquad \qquad t$$

$$\qquad \qquad \qquad \text{mol}$$

$$\qquad \qquad \qquad t$$

$$n_{CH_4}^H(t)$$

$$n_{CH_4}^H(t) = [n_{CH_4}^G(0) + n_{CH_4}^O(0)] - [n_{CH_4}^G(t) + n_{CH_4}^O(t)]$$

2

$$n_{CH_4}^G(0) \quad n_{CH_4}^O(0) \qquad \qquad \qquad 0 \qquad \qquad \qquad \text{mol}$$

$$0 \quad n_{CH_4}^G(t) \quad n_{CH_4}^O(t) \qquad \qquad \qquad t \qquad \qquad \qquad \text{mol}$$

$$0 \qquad \qquad \qquad t$$

$$n_{CH_4}^G = \frac{PV}{ZRT}$$

3

$$n_{CH_4}^G \qquad \qquad \qquad P \quad V \quad T \qquad \qquad \qquad R$$

$$8.3145 \text{ J/(mol K)} \quad Z \qquad \qquad \qquad PR$$

$$x = \frac{P}{H}$$

4

$$x \qquad \qquad \qquad P \qquad \qquad \qquad \text{MPa} \quad H \qquad \qquad \qquad \text{MPa}$$

$$\qquad \qquad \qquad x$$

$$x = \frac{n_{CH_4}^O}{n_{CH_4}^O + n_{oil}}$$

5

H

[34]

$$\ln H = 38 - \frac{1966}{T} - 4.9 \ln T$$

7

T

K

1.4.2 水合储气速率

10 s

8

-

t

9

$$r = \frac{dC_v}{dt} \approx \frac{\Delta C_v}{\Delta t'} \left[\text{L} / (\text{L} \cdot \text{min}) \right]$$

8

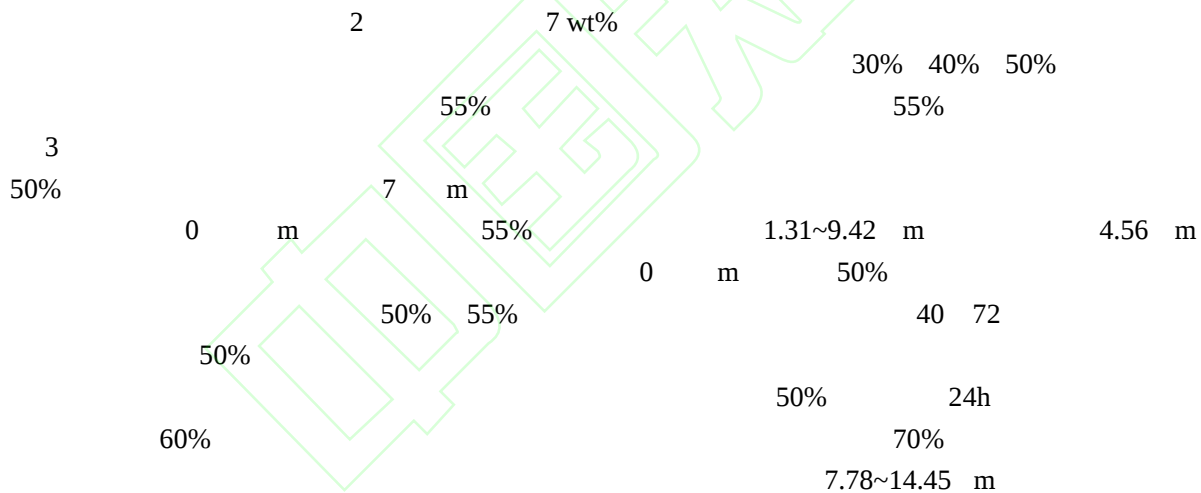
$$r_a = \frac{\Delta C_v}{\Delta t} \left[\text{L} / (\text{L} \cdot \text{min}) \right]$$

9

Δ t'

2

2.1



2

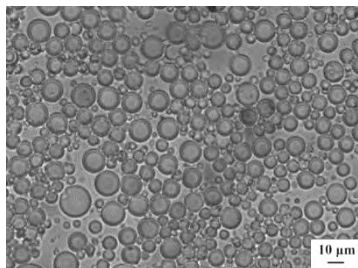
T=273.15 K P₀=7 MPa

7 wt%

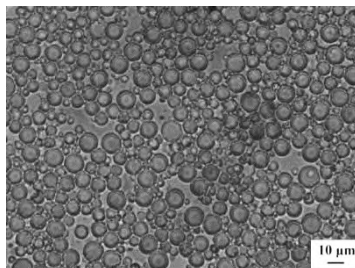
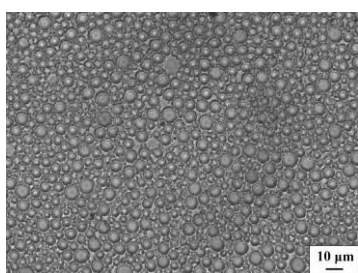
Table 2 Hydrate-based methane storage results of emulsions with different water-oil ratios

		mL	h	(L /L)	[L /(L h)]
1		50			
2	+7 wt%S80/T80	50			
3	30%	50			
4	40%	50			
5	50%	50			

6	55%	50	3.80	22.25	19.61
7	60%	50	0.19	17.26	49.56
8	70%	50	4.01	12.07	15.08



a 30/70 $d_g=10.12$ m

b 40/60 $d_g=8.54$ mc 50/50 $d_q=7.33$ m

d 55/45 $d_q=4.56$ m

e 60/40 $d_q=6.11$ m

f 70/30 $d_a =$ m /7.78~14.45
m)

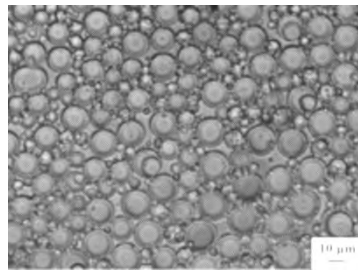
Fig. 3 Micromorphology of emulsions with different water-oil ratios

2.2

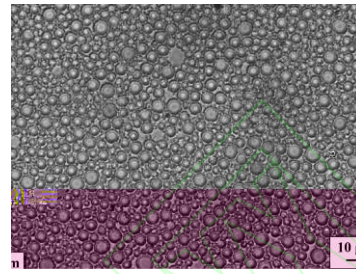
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Table 3 Hydrate-based methane storage results of emulsions with different emulsifier dosages

	wt%	mL	h	(L /L)	[L /(L h)]
1	1	50			
2	3	50			
3	5	50	1.05	25.31	25.13
4	7	50	3.80	22.25	19.61



a 1 wt% $d_a=11.38 \text{ m}$



b 3 wt% $d_a=9.28 \text{ m}$

c 5 wt% $d_a=5.33 \text{ m}$

d 7 wt% $d_a=4.56 \text{ m}$

Fig. 4 Micromorphology of emulsions with different emulsifier dosages

2.3

5 wt%

5

13.05 L / (L min)

1.45~8.82

55%
274.15 K

5 wt%
10:1

900 r/min

274.15 K

5.33

55%

118.41 L / L

6.08MPa

55%

Span80/Tween80

10:1

6

5wt%

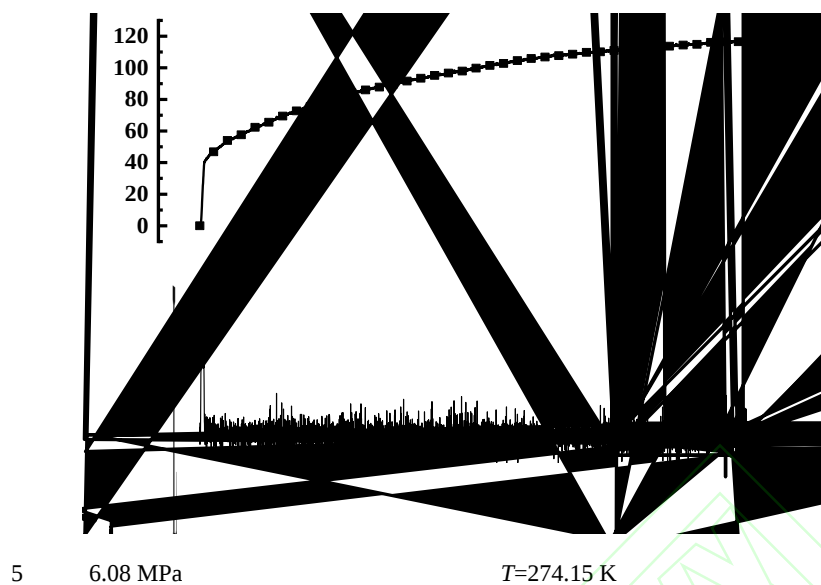
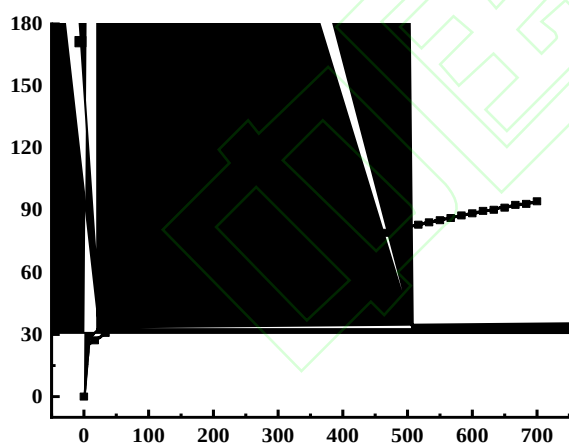
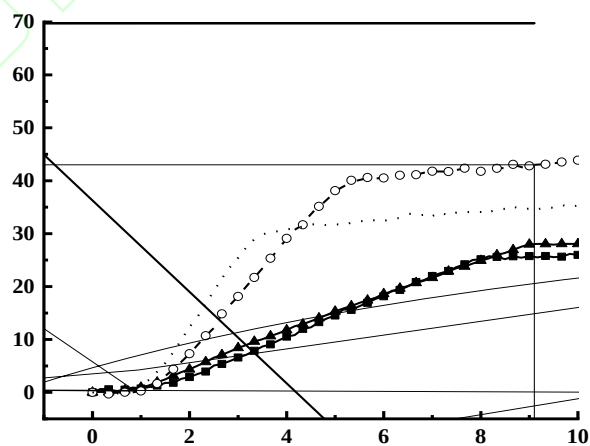


Fig. 5 Gas hydration kinetics of emulsions under 6.08 MPa

6
118.41 L /L
4.48 MPa
7.92 MPa
0.138 W/(m K)
94.30 L /L
83.59 L /L
6.93 MPa
4.48~6.08 MPa
6.08 MPa
116.84 L /L



a



b

6

T=274.15 K

900 r/min

Fig. 6 C_V -t curves of emulsion under different pressure conditions

2.4

7
274.15 K
6 MPa
300~1100
r/min
128.77 L /L
141.42 L /L
1100 r/min
68.65 L /L
300 r/min
700 r/min
700~1100 r/min

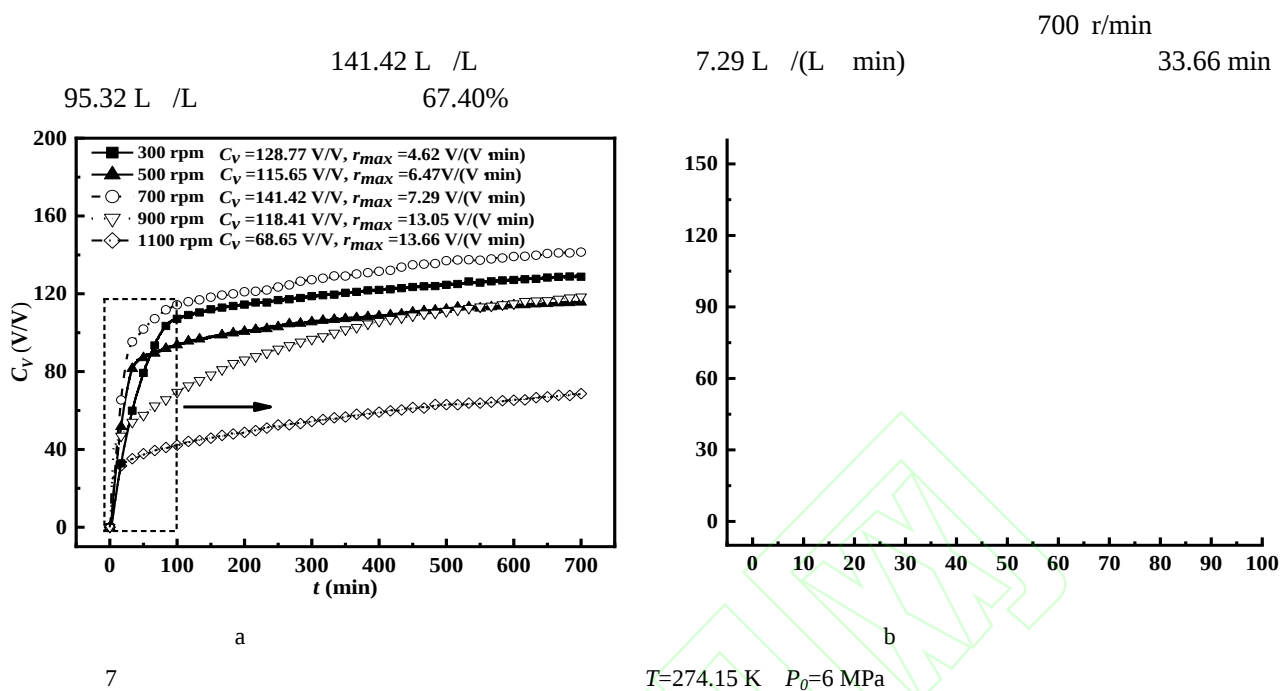


Fig. 7 C_V - t curves of emulsion in different stirring rate systems

2.5

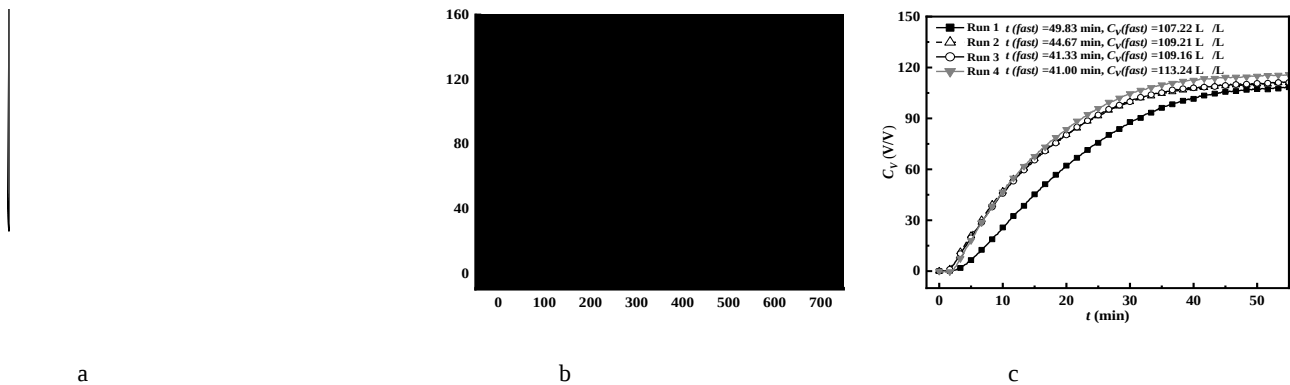
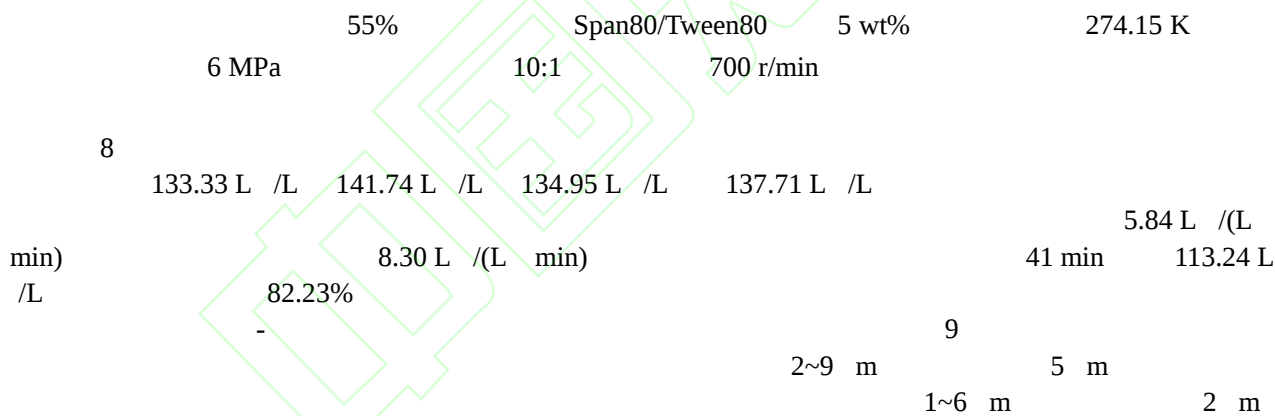


Fig. 8 C_V - t curves of cycling hydrate based methane storage in emulsion

a d_a 5 μm b d_a 2 μm

9 d_a

Fig. 9 Comparison of the micro-morphology of the emulsion before and after experiment

3

了

于

Span80/Tween80

5wt%

6

274.15 K

141.42 L /L

130V/V

10:1

6 MPa

700 r/min

5.84 L /L min)

8.30 L

5.33

55%

5

2

/(L min)

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